

Cambridge IGCSE[™]

CANDIDATE NAME					
CENTRE NUMBER			CANDIDATE NUMBER		

836961267

BIOLOGY 0610/61

Paper 6 Alternative to Practical

October/November 2022

1 hour

You must answer on the question paper.

No additional materials are needed.

INSTRUCTIONS

- Answer all questions.
- Use a black or dark blue pen. You may use an HB pencil for any diagrams or graphs.
- Write your name, centre number and candidate number in the boxes at the top of the page.
- Write your answer to each question in the space provided.
- Do not use an erasable pen or correction fluid.
- Do not write on any bar codes.
- You may use a calculator.
- You should show all your working and use appropriate units.

INFORMATION

- The total mark for this paper is 40.
- The number of marks for each question or part question is shown in brackets [].

This document has 12 pages. Any blank pages are indicated.

1 A student investigated the vitamin C concentration in three different health drinks: A, B and C.

The concentration of vitamin C can be estimated in a sample of health drink by testing it with iodine solution. The test involves adding drops of iodine solution to the health drink sample until the sample remains blue-black.

The greater the volume of iodine solution added, the higher the concentration of vitamin C in the health drink.

The student used this method:

- Step 1 Label three test-tubes A, B and C.
- Step 2 Use a syringe to add 1 cm³ of starch suspension to each of test-tubes **A**, **B** and **C**.
- Step 3 Use a clean syringe to add 3 cm³ of health drink **A** to test-tube **A**.
- Step 4 Use a clean syringe to add 3 cm³ of health drink **B** to test-tube **B**.
- Step 5 Use a clean syringe to add 3 cm³ of health drink **C** to test-tube **C**.
- Step 6 Use a dropping pipette to add one drop of iodine solution to test-tube **A** and shake the test-tube gently to mix the contents.
- Step 7 Repeat step 6, counting the number of drops added, until a blue-black colour appears and stays blue-black after mixing.
- Step 8 Record the total number of drops of iodine solution added to test-tube **A**.
- Step 9 Repeat steps 6 to 8 with test-tubes **B** and **C**. Refill the dropping pipette with more iodine solution as necessary.
- Step 10 Pour the contents of test-tubes **A**, **B** and **C** into the waste container and rinse the test-tubes in clean water. Repeat steps 2 to 9 to obtain another set of results.
- Step 11 Repeat step 10 to obtain a third set of results.

The student's results are shown in Fig. 1.1.

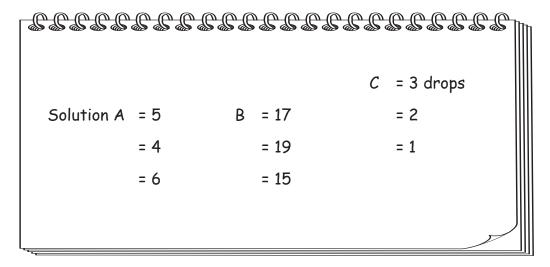


Fig. 1.1

Calculate the average number of drops of iodine solution used for each solution.

(a) (i) Prepare a table and record the results shown in Fig. 1.1.

Include these averages in your table.

	[4]
(ii)	Suggest a reason why the starch suspension was added to each of the test-tubes in this investigation.
	[1]
(iii)	State which health drink, A , B or C , has the highest concentration of vitamin C and give a reason for your answer.
(i)	
(iv)	Identify the variable that was measured (the dependent variable) in this investigation. [1]

		lain why it was important t					
(vi)		ntify one possible source of hod to reduce the effect o		ps 6 and 7 ar	nd suggest a	an improveme	ent to th
	erro	r					
	imp	rovement					
							[2
	ine so	olution until each concentr	ation of vitan	nin C solution	i remained i	Jiue-black.	
iodi The ado	e expe ded w	Plution until each concentrate or concentrate or concentrate as calculated for each cor shows the student's results.	e times and the contraction.	the average i			solutio
iodi The ado	e expe ded w	eriment was repeated thre as calculated for each cor I shows the student's resu	e times and incentration. Ilts. Table	the average	number of di	rops of iodine	solutio
iodi The add	e expe ded w	eriment was repeated thre as calculated for each cor	e times and incentration. Ilts. Table number	the average of the av	number of di	rops of iodine	solutio
iodi The add	e expe ded w	eriment was repeated thre as calculated for each cor I shows the student's resu percentage vitamin C concentration	e times and incentration. Ilts. Table	the average	number of di	on added average	solutio
iodi The ado	e expe ded w	eriment was repeated thre as calculated for each cor shows the student's resupercentage vitamin C	re times and incentration. Ilts. Table number trial 1	the average of the av	number of di	rops of iodine	solutio
iodi The add	e expe ded w	percentage vitamin C concentration 0.000	ncentration. Its. Table number trial 1	the average of the drops of in trial 2	odine solution trial 3	on added average	solutio
iodi The add	e expe ded w	percentage vitamin C concentration 0.000 0.025	ncentration. Its. Table number trial 1 1 3	e 1.1 of drops of identification of trial 2 1 3	odine solution trial 3	on added average 1 3	solutio
iodi The ado	e expe ded w	percentage vitamin C concentration 0.000 0.025 0.050	ncentration. Its. Table number trial 1 1 3 4	e 1.1 of drops of id trial 2 1 3 19	odine solution trial 3 1 3 6	on added average 1 3 5	solutio
iodi The	e expe ded w	percentage vitamin C concentration 0.000 0.025 0.100	number trial 1 3 4	of drops of intrial 2 1 3 19 10	odine solution trial 3 1 3 6 12	on added average 1 3 5 11	soluti

(ii)	State how the student dealt with the anomalous result when calculating the average value for the 0.050% vitamin C solution.
(iii)	Using the data in Table 1.1, plot a line graph on the grid of the percentage vitamin C
···· <i>)</i>	concentration against the average number of drops of iodine solution added.
	[4]
(iv)	The student was given a health drink, ${\bf D}$. It took seven drops of iodine solution to change it to a blue-black colour.
	Use the graph to estimate the vitamin C concentration in health drink D .
	On the graph, show how you estimated the vitamin C concentration.
	vitamin C concentration of D

escribe one way in which the method for testing a substance with Benedict's solution ffers from the method for testing a substance with biuret solution.	(v)
[1	
[Total: 19	

2 (a) Fig. 2.1 is a photograph of the largest butterfly in the world, the Queen Alexandra's birdwing butterfly.



magnification ×0.3

Fig. 2.1

(i)	Line EF represents the wingspan of the Queen Alexandra's birdwing butterfly.
	Measure the length of line EF on Fig. 2.1.

length of line **EF**......mm

Calculate the actual wingspan of the butterfly using the formula and your measurement.

$$magnification = \frac{length \ of \ line \ \textbf{EF}}{actual \ wingspan \ of \ the \ butterfly}$$

Give your answer to **two** significant figures.

Space for working.

 . mm
[3]

(ii) Fig. 2.2 shows one wing from the Queen Alexandra's birdwing butterfly.



Fig. 2.2

Draw a large diagram of the butterfly wing shown in Fig. 2.2.

(b) Fig. 2.3 is a photograph of a Queen Alexandra's birdwing caterpillar.



Fig. 2.3

Fig. 2.4 is a photograph of a monarch butterfly caterpillar.



Fig. 2.4

Identify **two** differences between the Queen Alexandra's birdwing caterpillar in Fig. 2.3 and the monarch butterfly caterpillar in Fig. 2.4.

1	
2	
	[2]

(c) Caterpillars turn into butterflies after pupating.

A scientist investigated the effect of temperature on the number of days taken for a caterpillar to become a butterfly.

The scientist collected 120 caterpillars of one species of butterfly. All the caterpillars were the same age.

Four glasshouses were prepared and each was maintained at a different temperature.

30 caterpillars were placed in each glasshouse. Each group of caterpillars was provided with the same mass of food.

The scientist observed the caterpillars in the glasshouses and recorded the number of caterpillars remaining at each temperature.

Table 2.1 shows the scientist's results at 30 days and 60 days.

Table 2.1

temperature/°C	number of caterpillars remaining at 30 days	number of caterpillars remaining at 60 days
15	26	25
20	9	2
25	5	0
30	24	22

(i)	State two variables that were kept constant in this investigation.
	1
	2
	[2
(ii)	Calculate the percentage change in the number of caterpillars remaining between 30 days and 60 days when the temperature was 15 °C.
	Give your answer to one decimal place.
	Space for working.

.....%

	(iii) Use the information in Table 2.1 to suggest the optimum (best) temperature for ch from a caterpillar into a butterfly.	anging
		°C [1]
	[То	otal: 15]
3	Plan an investigation to determine the effect of temperature on the rate of photosynthesis aquatic plant.	s in an
		[6]

BLANK PAGE

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced online in the Cambridge Assessment International Education Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download at www.cambridgeinternational.org after the live examination series.

Cambridge Assessment International Education is part of Cambridge Assessment. Cambridge Assessment is the brand name of the University of Cambridge Local Examinations Syndicate (UCLES), which is a department of the University of Cambridge.